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In Silico-Guided Identification of New Potent Inhibitors of Carbonic Anhydrases Expressed in *Vibrio cholerae*

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ABSTRACT: Carbonic anhydrases from *Vibrio cholerae* (VchCAs) play a significant role in bacterial pathophysiological processes. Therefore, their inhibition leads to a reduction of gene expression virulence and bacterial growth impairment. Herein, we report the first ligand-based pharmacophore model as a computational tool to study selective inhibitors of the β-class of VchCA. By a virtual screening on a collection of sulfonamides, we retrieved 9 compounds that were synthesized and evaluated for their inhibitory effects against VchCAβ as well as α- and γ-classes of VchCAs and selectivity over human ubiquitous isoforms hCA I and II. Notably, all tested compounds were active inhibitors of VchCAs. The N-(4-sulfamoylbenzyl)-[1,1'-biphenyl]-4-carboxamide (20e) stood out as the most exciting inhibitor toward the β-class ($K_i = 95.6$ nM), also showing a low affinity against the tested human isoforms. By applying docking procedures, we described the binding mode of the inhibitor 20e within the catalytic cavity of the modeled open conformation of VchCAβ.

KEYWORDS: Carbonic anhydrase inhibitors, Vibrio cholerae, ligand-based modeling, sulfonamides, molecular docking

he Gram-negative bacterium Vibrio cholerae (Vch) is the causative agent of cholera, a severe diarrheal disease that is endemic in various Southeast Asian and African countries as well as regions of South America. 1,2 This pathology can lead to severe dehydration, metabolic acidosis, and death in the absence of therapeutic intervention. It is well-known that Vch colonizes gastro-intestinal lumen and causes pathological effects by producing virulence factors related to transcriptional regulator ToxT.3-5 Consequently, an emerging challenge is to fight cholera by using antivirulence drug candidates in place of antibiotics. It has been established that the ToxT activation is regulated by ion bicarbonate (HCO₃⁻) as intestinal pH buffer secreted by epithelial cells.⁷ The bicarbonate production is mediated by carbonic anhydrases (CAs, EC 4.2.1.1) that are metalloenzymes catalyzing the reversible hydration of CO₂. CAs are a superfamily of enzymes belonging to several classes $(\alpha$ -, β -, γ -, δ -, ζ -, ε -, θ -, and ι -classes) that are diffused in vertebrates, protozoa, algae, and bacteria. 8,9 Specifically, the Vch genome encodes α -, β -, and γ -classes named VchCA α , VchCA β , and VchCA γ , which share a low structural homology

with each other. The α - and β -CAs are Zn(II) metalloenzymes, whereas γ -CA classes employ Fe(II) in the catalytic site, even if they are also active with Zn(II) or Co(II) metal ions. In detail, the metal ion is coordinated by three His residues in the α - and γ -classes and one His and two Cys residues in the β -class. The coordinate is the β -class.

The ability to inhibit VchCA isozymes has been demonstrated by a large series of compounds bearing a zinc binder group (ZBG). Among them, acetazolamide (AAZ) and ethoxzolamide (EZA) (Figure 1) proved to significantly inhibit VchCA isozymes. 9,11 Furthermore, EZA decreases the bicarbonate-mediated virulence gene expression and reduces

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$$SO_2NH_2$$
 SO_2NH_2
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 SO_2NH_2

Figure 1. Chemical structures of well-known VchCA inhibitors: acetazolamide (AAZ) and ethoxzolamide (EZA).

the growth of pathogen; this latter evidence paved the way for the development of VchCA inhibitors as potential therapeutics for treatment of cholera. ^{9,12,13}

The design of selective inhibitors targeting the medium/small cavity of VchCA isozymes is a very intriguing challenge. Actually, the best active inhibitors displayed good affinity in the low nanomolar range. $^{11,12,14-21}$ However, they generally display low selectivity over human off-target $\alpha\text{-class}$ CA isoforms (hCA I and hCA II), thus reducing their potential therapeutic interest in humans. The CA inhibitors possess the ZBG linked to a lipophilic cap-group through a suitable "spacer". Apart from that crucial ZBG moiety, structural studies have suggested that the cap-group can establish relevant contacts with the rim of the CA cavity thus controlling CA isozyme selectivity.

Seeking selective inhibitors targeting the medium/small cavity of VchCA, we focused our interest on β -CA classes which exert catalytic activity as a crucial event for bacterial survival. It has been established that β -CAs are structurally unique from human CAs, so that it has been proposed that the selective β -CA inhibitors might be promising innovative antibacterial agents. 22,23 To better understand the binding mode of VchCA inhibitors, we have previously investigated the pose of prototype AAZ bound to the hypothetical cavity located in the interface of a dimeric VchCA β , that displays a tetrameric composition as a dimer of dimers. In detail, we have modeled the open active site on the basis of the cocrystal structure of AAZ in complex with β -CA from the green algae Coccomyxa (PDB Code: 3UCJ).²⁴ This in silico study suggested that the deprotonated sulfonamide moiety is anchored to zinc ion, which is coordinated by residues Cys42, His98, and Cys101 (chain A, blue in Figure 2); a hydrogen bonding interaction was found between the oxygen atom of Gly102 (chain A) and the exocyclic nitrogen atom of

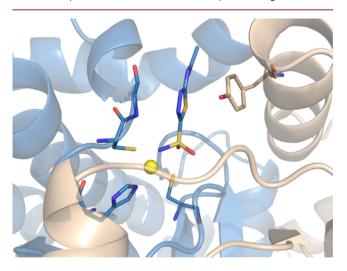


Figure 2. Binding site analysis of the modeled dimeric VchCA β open cavity bound to acetazolamide (AAZ).

the acetamide moiety; finally, the nitrogen atom of the thiadiazole ring establishes H-bond contact with the hydroxyl group of Tyr83 (chain B, wheat in Figure 2), for which a π/π stacking with the thiadiazole nucleus might reinforce the binding within the catalytic site.

Continuing our efforts aimed to the identification of selective CA inhibitors, $^{14,25-36}$ in this work we report a computationally driven design of small molecules which possess the minimal structural requirements to occupy the tight cleft of the β -CA cavity and establish favorable contacts with crucial residues of VchCA isozymes, thus anchoring the catalytic site as well as hydrophobic/hydrophilic walls of the CA cavity.

Our study began with a ligand-based strategy to obtain a three-dimensional pharmacophore model, that was validated to establish its robustness as a valuable tool to identify a compound having affinity to the β -class of VchCA. Then, a structure-based virtual screening of 3D-databases allowed us to select hypothetical drug-like sulfonamides able to establish interaction with enzymatic cavity. Finally, nine compounds were synthesized and screened to establish the reliability of our computational study and reach new information about the SAR for selective and potent VchCA β inhibitors over hCA I and hCA II. Then, the hypothetical binding pose was suggested by docking studies. The following sections describe the step-by-step procedure to achieve our goal.

We initially constructed our pharmacophore model for CA inhibitors targeting VchCA β isozymes. To assemble the data set, we selected from the literature ^{14,21} 19 known inhibitors (compounds 1-19, Figure 3) that possess the RSO₂NHR chemical moiety as well-established ZBG; to guarantee the best reliability of the pharmacophore hypotheses, we selected a homogeneous series of inhibitors that have been assayed by employing the stopped-flow carbon dioxide hydrase assay; compounds 1–19 displayed K_i values ranging from 68 to 6000 nM as active compounds. Then, the data set compounds were distributed in two subsets of compounds: the training set (Figure 3, compounds 1-9) and test set (Figure 3, compounds **10−19**). By employing the above-mentioned two data sets of sulfonamides, a collection of ten pharmacophore models was generated by LigandScout;³⁷ then, the pharmacophore hypothesis with the best score value (72.7165) was considered from the 10 generated models. To validate this pharmacophore model, we established its discrimination power by considering the enrichment factor and the area under the curve (AUC) of the receiver operating characteristic (ROC) curve (for details see Figure 6 in the Supporting Information); the model displays a preference for active compounds with an AUC value of 0.97 and EF of 11.5.

As shown in Figure 4A the best pharmacophore model consists of one aromatic ring feature (blue), one hydrophobic feature (yellow), three hydrogen bond acceptors (red), two hydrogen bond donors (green), one negative ionizable (red star), and 31 exclusion volumes (gray).

The alignment of the 3D coordinates of the active inhibitor **AAZ** (K_i value of 451 nM) onto the pharmacophore (see Figure 4B) highlighted that the deprotonated form of sulfonamide moiety (RSO₂NH $^-$) is defined by two oxygen atoms as hydrogen bond acceptors as well as one nitrogen atom corresponding to a hydrogen bond donor or a negative ionizable group; in addition, the heteroaromatic ring describes a hydrophobic/aromatic ring feature. Furthermore, the other hydrogen bond donor feature corresponds to the nitrogen

Figure 3. Chemical structures of compounds 1-19.

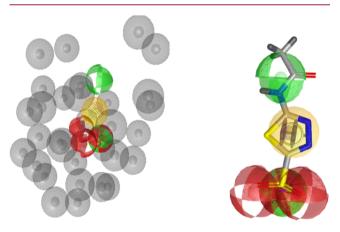


Figure 4. (A) Best pharmacophore model: one aromatic ring feature (blue) overlapped with the hydrophobic feature (yellow), three hydrogen bond acceptors (red), two hydrogen bond donors (green), one negative ionizable (red star), and 31 exclusion volumes (gray). (B) AAZ mapped into the pharmacophore model.

atom of the amide portion. Notably, the above-described features resulted in good agreement with previously reported contacts found for **AAZ** docked into our "modeled" open conformation of VchCA β catalytic cavity (cf. Figure 2).

Encouraged by this strong matching between the pharmacophore model displayed in Figure 4A and docking results for AAZ, we employed this plausible model to carry out a virtual screening (VS) in order to identify new of VchCA β inhibitors. The second step of our computational study involved the construction of a plausible database of VchCA β inhibitors through the collection of para-benzenesulfonamide derivatives retrieved from the SciFinder chemical database (https:// scifinder.cas.org). We employed specific filters to select both drug-likeness and commercially available compounds; therefore, we collected 8,208 molecules that were screened by our best pharmacophore model, thus obtaining 661 compounds. Among them we selected 118 molecules having fit-score values greater than 72.86. By visual inspection we focused our interest on 40 compounds, that were docked into our modeled β -CA cavity by Gold software.³⁸ Therefore, docking analysis afforded 9 sulfonamides that were selected on the basis of a very simple synthetic procedure to obtain them.

As depicted in Figure 5 the selected compounds 20a-i are characterized by the canonical ZBG linked to the lipophilic

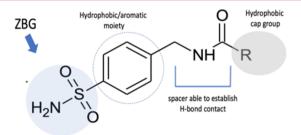


Figure 5. Schematic representation of structural moieties shared by sulfonamides 20a-i

cap-group by an amide spacer as a crucial motif to bind the VchCA β catalytic cavity through the requested H-bond donor group; additionally, the cap group might furnish a selective interaction with VchCAs over other CA classes.

By coupling the 4-aminomethylbenzenesulfonamide (21) with commercially available carboxylic acids or aroyl chlorides, we obtained in good yields the small series of desired *N*-(4-sulfamoylbenzyl)amide derivatives 20a-i (Scheme 1). The chemical characterization of compounds 20a-i was supported by elemental analyses and ¹H and ¹³C NMR spectroscopic measurements.

Scheme 1. Synthetic Route for Desired N-(4-Sulfamoylbenzyl)amide Derivatives 20a-i^a

^aReagents and conditions: (i) (A) RCOCl, DIPEA, DCM/DMF (2:1, v/v), MW, 25 °C, 10 min; (B) RCO₂H, HBTU, DIPEA, DCM/DMF (2:1, v/v), MW, 25 °C, 25 min.

The N-(4-sulfamoylbenzyl)amide derivatives 20a-i were assayed for their inhibitory activity against VchCA α , - β , and - γ by means of a stopped-flow carbon dioxide hydrase assay. The obtained results are summarized in Table 1 and compared with K_i values of AAZ as reference compound. For comparative purposes the inhibitory profiles against the ubiquitous hCA I and hCA II are reported in Table 1.

Table 1. Inhibitory Effects against VchCAα, VchCA-β, VchCA-γ, hCA I, and hCA II of Compounds 20a—i and Reference Compound Acetazolamide (AAZ)

	K_i (nM)				
	VchCAα	VchCβ	VchCAγ	hCA I	hCA II
20a	45.0	6442.0	56.1	60.7	3.3
20b	9.1	626.7	250.8	65.9	5.1
20c	8.8	3596.0	722.4	77.8	31.6
20d	18.1	179.2	98.4	95.0	63.0
20e	11.6	95.6	174.6	2113.0	919.7
20f	12.1	586.1	657.4	98.3	54.4
20g	6.2	553.9	593.0	269.3	26.3
20h	10.0	200.4	775.0	44.2	83.8
20i	7.7	538.5	79.6	571.5	69.5
AAZ	6.8	451.0	470.0	250.0	12.1

 $[^]a\mathrm{Errors}$ in the range of $\pm 10\%$ of the reported value, from 3 different assays.

All computationally inspired compounds 20a-i affected the carbon dioxide hydrase activity of VchCA classes showing Ki values ranging from 6.2 to 6442 nM. The screening toward VchCA α evidenced that the R substituent did not significantly affect the inhibitory potency of tested compounds 20a-i. Notably, the 1-(biphenyl-4-yl)-substituted compound 20e was the most active VchCA β inhibitor (K_i value of 95.6 nM). Compounds 20b, 20f, 20g, and 20i were about 5-fold less active VchCA β inhibitors when compared with analogue **20e** (K_i value of 95.6 nM). The presence of m-tolyl or 2,5chlorophenyl as hydrophobic group was critical for VchCA β affinity and dramatically reduced the activity toward VchCA β of compounds 20a and 20c, respectively. On the contrary, compounds 20d and 20h were more active when compared with 20a and 20c, whereas they were weakly active with respect to the most interesting inhibitor 20e. All these data evidenced how changes in the size and/or shape of the hydrophobic fragment can impact the inhibitory profile toward VchCA β . The inhibitory trend toward VchCA γ revealed that compounds 20a, 20d, and 20i were active at low nanomolar concentration; the remaining compounds of the series resulted weak inhibitors. Taken together these data confirmed that VchCA α is more able to accommodate the various Rsubstituents compared to the other tested isozymes VchCA- β and VchCA-\gamma. Overall, the most relevant result was the identification of compound 20e as a potent and selective VchCA- β inhibitor that displayed very low affinity toward human CA isoforms hCA I and hCA II (K; values of 2113.0 and 919.7 nM, respectively).

A further step of our study was to analyze the hypothetical orientations into the VchCA β cavity for synthesized compounds by docking analysis, that were performed by means of Gold Software.³⁸ In detail, we used the crystal structure of the dimeric (chains A and B) VchCA β retrieved from the Protein Data Bank (PDB Code: 5CXK),¹⁰ that has been "modeled" in open conformation on the basis of β -CA from the green algae

Coccomyxa²⁴ (PDB Code: 3UCJ) as previously reported and shown in Figure 2 for AAZ.¹⁴ The docking results confirmed that N-(4-sulfamoylbenzyl)amide derivatives 20a-i adopted the canonical orientation of sulfonamide-based CAIs for which for the deprotonated form of the sulfonamide moiety is anchored to the zinc ion coordinated by residues Cys42, His98, and Cys101 (chain A, colored in blue). As expected, the aromatic ring of the benzenesulfonamide portion is stabilized through a π - π interaction with Tyr83 (chain B, colored in wheat). In addition, the -NH- group of the amide spacer establishes H-bond interaction with the oxygen atom of the Gly102 backbone. Our studies suggested that the cap-group might be involved in a network of interactions with a cluster of residues Thr105, Ala106, Ala139, and Ile108 lining the hydrophobic subpocket along the rim of chain A. The network of above-mentioned interactions is displayed in Figure 6 for the most active inhibitor N-(4-sulfamoylbenzyl)biphenyl-4carboxamide (20e, K_i value of 96.5 nM).

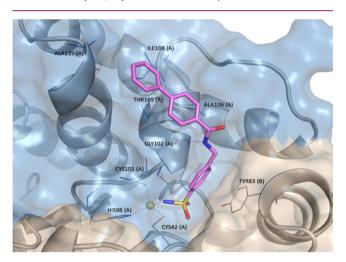


Figure 6. Plausible binding mode of **20e** into our "modeled" open conformation of VchCA β . Dark gray dashed lines represent hydrogen bond interaction. Zinc ion is depicted as a yellow sphere. The interactions were examined by using LigandScout software.³⁷ The images were created by means of PyMOL software (https://pymol.org).

In conclusion a ligand based virtual screening strategy led to the identification of compound **20e** as active VchCA- β inhibitor (K_i value of 95.6 nM) that combined high affinity with a surprising selectivity over the human off-target isoform. The screening and docking efforts established that this compound might be a promising lead compound for further biological studies, HTS screening, and structural optimization aimed to the identification of novel anti-infective agents characterized by a peculiar mechanism of action, in order to overcome the global threat of antibiotic resistance.

ASSOCIATED CONTENT

5 Supporting Information

The Supporting Information is available free of charge at https://pubs.acs.org/doi/10.1021/acsmedchemlett.0c00417.

Experimental details (synthetic experimental details, analytical and ¹H and ¹³C spectral data, biochemical assays, and computational studies) (PDF)

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Author Contributions

All authors contributed to data interpretation. The manuscript was written through contributions of all authors. All authors have given approval to the final version of the manuscript.

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Notes

The authors declare no competing financial interest.

ABBREVIATIONS

CAs, carbonic anhydrases; Vch, *Vibrio cholerae*; AAZ, acetazolamide; EZA, ethoxzolamide; HCT, hydrochlorothiazide; BZA, benzolamide; SAC, saccharine; BRZ, brinzolamide; SLP, sulpiride; SLT, sulthiame

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